

Fig. 1. Perspective view of the title molecule (ORTEPII; Johnson, 1976) with $30 \%$ probability ellipsoids.
$\mathrm{O}(2)-\mathrm{C}(7)$ bond $\quad\left[D_{s}(\mathrm{~S})=0.0081\right.$ (21) (Nardelli, 1983)]. Such a conformation is quite similar to that reported by Uang, Liu \& Wang (1990) for an oxathiolanone bornane derivative and is also in agreement with the structure suggested for several oxathiolanones by Farines (1973) and Pihlaja, Nikkila, Neuvonen \& Keskinen (1976) from NMR data. The two quite different $\mathrm{S}-\mathrm{C}$ distances observed in this ring can be justified taking into account the different values between the angles $\mathrm{S}-\mathrm{C}(8)-\mathrm{C}(7) \quad\left[107.5(4)^{\circ}\right] \quad$ and $\quad \mathrm{S}-\mathrm{C}(1)-\mathrm{O}(2)$ [104.7 (3) ${ }^{\circ}$. Therefore, the hybridization (and hence the overlapping ability and electronegativity) of both $\mathrm{C}(8)$ and $\mathrm{C}(1)$ atoms will vary with the internal angle and there is no single rule for predicting the results. Finally, the ${ }^{1} \mathrm{H}$ NMR spectra for this compound clearly show the anisotropic influence of the $\mathrm{S}-\mathrm{O}(1)$ bond on the two protons bonded to the $\mathrm{C}(8)$ atom, which are each located at different sides of the ring.

Obviously, the position of the $\mathrm{S}=\mathrm{O}(1)$ bond, almost perpendicular to this ring, destroys their magnetic equivalence, so they then appear as an $A B$ spin system ( $J=17.7 \mathrm{~Hz}$ ).

This research was supported by CICYT (grant PB89-0417).

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Acta Cryst. (1992). C48, 1525-1527

# Structure of Tris(phenylthio)phosphine Sulfide 

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(Received 2 August 1991; accepted 17 December 1991)


#### Abstract

C}_{18} \mathrm{H}_{15} \mathrm{PS}_{4}, M_{r}=390.5\), monoclinic, $P 2_{1} / n$, $a=10.1406$ (5), $b=10.1948$ (6), $c=18.870$ (1) $\AA, \beta$ $=97.253(2)^{\circ}, \quad V=1935.2(3) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.34 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda($ Mo $K \alpha)=0.71069 \AA \quad$ (graphite monochromator), $\mu=5.49 \mathrm{~cm}^{-1}, F(000)=808, T=$ $297 \mathrm{~K}, R=0.045$ for 2507 observed reflections with $I$ $>3 \sigma(I)$. Distances: $\mathrm{P}=\mathrm{S} 1.899(1), \mathrm{P}-\mathrm{S}$ (av.) 2.102 (2) $\AA$; angles (av.): $\mathrm{S}=\mathrm{P}-\mathrm{S} 117.1$ (3), $\mathrm{S}-\mathrm{P}-\mathrm{S}$ $109.9(2)^{\circ}$. These are the first crystallographic data for the $\operatorname{SP}(\mathrm{SCR})_{3}$ unit.


Experimental. $\mathrm{SPCl}_{3}$ (Aldrich) was slowly added to $\mathrm{NaSC}_{6} \mathrm{H}_{5}$ (Shibao, Keder \& Eckert, 1990) in thiophenol under an $\mathrm{N}_{2}$ atmosphere at 295 K and allowed to react. The mixture was filtered and colorless crystals (plates) formed by slow solvent evaporation at room temperature. X-ray measurements were performed on a crystal $0.58 \times 0.71 \times 0.16 \mathrm{~mm}$ using a Huber (Crystal-Logic automated) four-circle diffractometer with $\theta / 2 \theta$ scan mode to a maximum $2 \theta$ of $50^{\circ}$ at a scan speed of $6^{\circ} \mathrm{min}^{-1}$. Lattice
parameters were determined from 35 reflections in the range $6.0<2 \theta<23.1^{\circ}$. Analytical absorption correction; maximum/minimum transmission factors were $0.887 / 0.711$ (Busing \& Levy, 1957). Maximum $\sin \theta / \lambda$ of $0.595 \AA^{-1} ; h, k, l$ range, $0-12,0-12$, -22-22. Three standard reflections were measured after every 97 reflections and showed no significant variation in intensity $\left[\left(I_{\max }-I_{\min }\right) / I_{\mathrm{av}}=0.027\right] .3823$ reflections were measured, 3606 were unique ( $R_{\text {int }}=$ $2.3 \%$ ) and 1099 were unobserved with $I<3 \sigma(I)$. The direct methods program SHELXS86 (Sheldrick, 1985) was used in solving the structure; $P$ and $S$ atoms located; C atoms were located by full-matrix least-squares refinement and difference Fourier syntheses; P, S and C positions and anisotropic thermal parameters were refined; $\quad \sum w\left|\left|F_{o}\right|-\left|F_{c}\right|^{2}\right.$ minimized where $w=1 /\left[\sigma\left(F_{o}\right)\right]^{2}$. H-atom positions were calculated ( $\mathrm{C}-\mathrm{H}$ bond length $1.0 \AA$ ) and included as fixed contributors with isotropic thermal parameters fixed to $5.0 \AA^{2} .208$ parameters refined; $R=0.045$, $w R=0.062, S=1.93 ;(\Delta / \sigma)_{\text {max }} 0.012 ; \Delta \rho_{\text {max }} / /_{\text {min }}=$ $0.46 /-0.44 \mathrm{e} \AA^{-3}$. Scatering factors were taken from International Tables for X-ray Crystallography (1974, Vol. IV). The UCLA Crystallographic Program Package (Strouse, 1985) was used throughout.


Fig. 1. ORTEPII (Johnson, 1976) drawing (50\% probability ellipsoids) of $\mathrm{SP}\left(\mathrm{SC}_{6} \mathrm{H}_{5}\right)_{3}$.


Fig. 2. Packing diagram ( $a$ axis towards viewer, $c$ axis horizontal).

Table 1. Atomic coordinates and equivalent isotropic temperature factors $\left(\AA^{2} \times 10^{4}\right)$ with e.s.d.'s in parentheses

| $U_{\text {eq }}=(1 / 3) \sum_{i} \sum_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i}, \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| $\mathrm{P}(1)$ | 0.12174 (9) | 0.44007 (9) | 0.33194 (5) | 536 (6) |
| S(1) | 0.2208 (1) | 0.5388 (1) | 0.27163 (5) | 666 (7) |
| S(2) | 0.0345 (1) | 0.26403 (9) | 0.29126 (5) | 669 (7) |
| S(3) | -0.04433 (9) | 0.5319 (1) | 0.36620 (6) | 684 (7) |
| S(4) | 0.2246 (1) | 0.3751 (1) | 0.42937 (5) | 697 (7) |
| C(21) | -0.0232 (4) | 0.3086 (3) | 0.2008 (2) | 582 (23) |
| C(22) | -0.1558 (4) | 0.3373 (4) | 0.1818 (2) | 725 (28) |
| C(23) | -0.2003 (4) | 0.3667 (5) | 0.1109 (2) | 835 (32) |
| C(24) | -0.1140 (5) | 0.3701 (4) | 0.0606 (2) | 773 (30) |
| C(25) | 0.0181 (5) | 0.3413 (4) | 0.0801 (2) | 763 (30) |
| C(26) | 0.0654 (4) | 0.3102 (4) | 0.1502 (2) | 696 (27) |
| C(31) | 0.0214 (3) | 0.6900 (3) | 0.3895 (2) | 578 (23) |
| C(32) | 0.0639 (4) | 0.7198 (5) | 0.4601 (2) | 769 (31) |
| C(33) | 0.1077 (5) | 0.8459 (6) | 0.4778 (3) | 1008 (42) |
| C(34) | 0.1053 (5) | 0.9394 (5) | 0.4268 (4) | 1017 (43) |
| C(35) | 0.0649 (5) | 0.9100 (4) | 0.3570 (3) | 906 (38) |
| C(36) | 0.0216 (4) | 0.7849 (4) | 0.3375 (2) | 693 (27) |
| C(41) | 0.3860 (3) | 0.3513 (4) | 0.4038 (2) | 572 (22) |
| C(42) | 0.4749 (4) | 0.4550 (4) | 0.4108 (2) | 760 (29) |
| C(43) | 0.6019 (5) | 0.4352 (6) | 0.3936 (3) | 948 (38) |
| C(44) | 0.6401 (4) | 0.3169 (6) | 0.3702 (3) | 922 (37) |
| C(45) | 0.5504 (5) | 0.2140 (5) | 0.3631 (2) | 851 (34) |
| C(46) | 0.4229 (4) | 0.2307 (4) | 0.3801 (2) | 681 (27) |

Table 2. Selected bond lengths $(\AA)$ and angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses

| $\mathrm{P}(1)-\mathrm{S}(1)$ | $1.899(1)$ | $\mathrm{S}(2)-\mathrm{C}(21)$ | $1.793(4)$ |
| :--- | :---: | :--- | :--- |
| $\mathrm{P}(1)-\mathrm{S}(3)$ | $2.099(1)$ | $\mathrm{S}(3)-\mathrm{C}(31)$ | $1.778(4)$ |
| $\mathrm{P}(1)-\mathrm{S}(2)$ | $2.103(1)$ | $\mathrm{S}(4)-\mathrm{C}(41)$ | $1.780(4)$ |
| $\mathrm{P}(1)-\mathrm{S}(4)$ | $2.103(1)$ |  |  |
| $\mathrm{S}(1)-\mathrm{P}(1)-\mathrm{S}(2)$ | $117.66(6)$ | $\mathrm{S}(3)-\mathrm{P}(1)-\mathrm{S}(4)$ | $101.50(6)$ |
| $\mathrm{S}(1)-\mathrm{P}(1)-\mathrm{S}(3)$ | $117.02(6)$ | $\mathrm{C}(21)-\mathrm{S}(2)-\mathrm{P}(1)$ | $101.7(1)$ |
| $\mathrm{S}(1)-\mathrm{P}(1)-\mathrm{S}(4)$ | $116.70(6)$ | $\mathrm{C}(31)-\mathrm{S}(3)-\mathrm{P}(1)$ | $101.0(1)$ |
| $\mathrm{S}(2)-\mathrm{P}(1)-\mathrm{S}(3)$ | $100.16(6)$ | $\mathrm{C}(41)-\mathrm{S}(4)-\mathrm{P}(1)$ | $100.2(1)$ |
| $\mathrm{S}(2)-\mathrm{P}(1)-\mathrm{S}(4)$ | $100.89(6)$ |  |  |

The program PLOTMD (Luo, Ammon \& Gilliland, 1989) was used to modify the labels of the ORTEP drawing (Johnson, 1976) displayed in Fig. 1. Positional parameters and isotropic temperature factors are listed in Table 1; bond lengths and angles are listed in Table 2.* In Fig. 2, the crystal packing is displayed.

Related literature. The structure of $\left(\mathrm{Me}_{3} \mathrm{SnS}\right)_{3} \mathrm{PS}$ (Shihada \& Weller, 1988), possessing a strain-free $\mathrm{SP}(\mathrm{SR})_{3}$ unit, has been reported. The terminal $\mathrm{P}-\mathrm{S}$ bond length in that compound [1.948 (2) $\AA$ ] is substantially longer than the one found here [1.899 (1) $\AA$ ].

[^0]Financial support of this research by NSF grant \# DMR 89-13738 is gratefully acknowledged.

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Acta Cryst. (1992). C48, 1527-1528

# Structure of 7-Acetyl-2,5,9-trinitro-2,5,7,9-tetraazabicyclo[4.3.0|nonan-8-one 

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(Received 19 August 1991; accepted 17 December 1991)


#### Abstract

C}_{7} \mathrm{H}_{9} \mathrm{~N}_{7} \mathrm{O}_{8}, M_{r}=319.2\), monoclinic, $P 2_{1} / n$, $a=10.615$ (1),$\quad b=9.115$ (2), $c=12.948$ (2) $\AA, \quad \beta=$ $108.57(1)^{\circ}, \quad V=1187.6(3) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.785 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{Cu} K \alpha)=1.54184 \AA, \quad \mu=$ $1.38 \mathrm{~mm}^{-1}, F(000)=656, T=295 \mathrm{~K}$, final $R=$ $0.039, w R=0.044$ for 1327 independent observed reflections. The five-membered ring has a normal envelope conformation while the six-membered ring has adopted a twisted boat conformation. Of the three nitroamine groups one is planar and the other two are pyramidal with $\mathrm{N}-\mathrm{N}$ to $\mathrm{C}-\mathrm{N}-\mathrm{C}$ angles of $0.5,43.4$ and $44.8^{\circ}$ respectively.


Experimental. A clear colorless prism $0.15 \times 0.20 \times$ 0.26 mm data crystal was provided by Dr Clifford L. Coon of Livermore National Laboratory, California. An automated Siemens $R 3 \mathrm{~m} / V$ diffractometer with incident beam monochromator was used for data collection. 25 centered reflections within $20.0 \leq 2 \theta \leq$ $77.0^{\circ}$ were used for determining lattice parameters. $[\sin (\theta) / \lambda]_{\max }=0.54 \AA^{-1}$; range of $h k l:-11 \leq h \leq 10$ $-9 \leq k \leq 2,0 \leq l \leq 13$. Standards 200, 020 and 002, monitored every 97 reflections, showed random variation of $2.5 \%$ over data collection: $\theta / 2 \theta$ scan mode, scan width $\left[2 \theta\left(K \alpha_{1}\right)-1.0\right]$ to $\left[2 \theta\left(K \alpha_{2}\right)+\right.$ $1.0]^{\circ}, \omega$-scan rate a function of count rate ( $7.0^{\circ} \mathrm{min}^{-1}$ minimum, $15.0^{\circ} \mathrm{min}^{-1}$ maximum), 1808 reflections measured, 1545 unique, $R_{\text {int }}=1.2 \%, 1327$ observed with $F_{o}>3 \sigma\left(F_{o}\right)$. Data were corrected for Lorentz and polarization effects.

The structure solution, by direct methods, and the full-matrix least-squares refinement used programs in SHELXTL80 (Sheldrick, 1980). $\quad \sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}$ was minimized where $w=1 /\left[\sigma^{2}\left(\left|F_{o}\right|\right)+g\left(F_{o}\right)^{2}\right], g=$
0.000225. Secondary-extinction parameter $p=$ $0.004(1) \quad$ in $\quad F_{c}^{*}=F_{c} /\left[1.0+0.002(p) F_{o}{ }^{2} / \sin (2 \theta)\right]^{0.25}$. 236 parameters were refined: atomic coordinates for all atoms, anisotropic thermal parameters for all non- H atoms and isotropic thermal parameters for H atoms. $(\Delta / \sigma)_{\max }=0.12$, ratio of observations to parameters $=5.6: 1, R=0.039, w R=0.044, S=1.39$ ( $R=0.046$ for all data). Final difference Fourier excursions were 0.18 and -0.20 e $\AA^{-3}$. Atomic scattering factors were taken from International Tables for X-ray Crystallography (1974, Vol. IV). $\dagger$ Atom numbering for Tables 1 (atom coordinates) and 2 (bond distances and angles) follows that shown in Fig. 1.
$\dagger$ Lists of structure factors, anisotropic thermal parameters and H -atom coordinates have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54969 ( 9 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England. [CIF reference: HH0573]


Fig. 1. A thermal ellipsoid plot of the title compound with ellipsoids drawn at the $20 \%$ probability level.
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[^0]:    * Lists of structure factors, anisotropic thermal parameters, H -atom coordinates and bond lengths and angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54964 ( 20 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: HH0567]

